Bisphosphorylation of a *vic*-Diol using a Phosphite Approach: Synthesis of *myo*-Inositol 4,5-Bisphosphate

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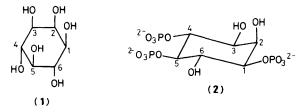
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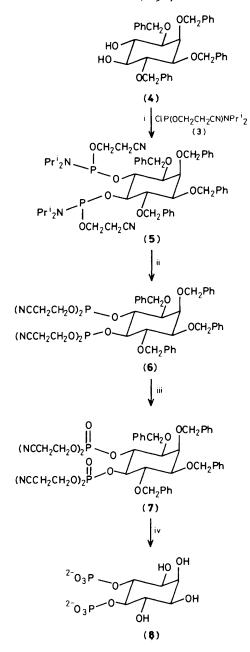
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(\pm)-*myo*-Inositol 4,5-bisphosphate was synthesized using a phosphite chemistry approach; bisphosphorylation was confirmed by the observation of the first ⁵J_{PP} coupling in a bisphosphite triester intermediate.

The recent discovery that a phosphate ester of *myo*-inositol (1), *D-myo*-inositol 1,4,5-trisphosphate (2),^{1,2} generated by receptor-stimulated phosphodiesteratic hydrolysis of phosphatidylinositol 4,5-bisphosphate, acts as the probable intracellular second messenger for calcium mobilisation, has created a dramatic upsurge of interest in inositol phosphates. *D-myo*-Inositol 1,3,4-trisphosphate³ and 1,3,4,5-tetrakisphosphate^{4,5} have also been discovered, but their biological roles are as yet unclear. Isolation of such biophosphates from natural sources is difficult and does not yield large quantities of material.⁶ Moreover, the need for radioactive derivatives and analogues of these molecules must be addressed by chemical synthesis.

Recently, the first chemical synthesis of (2) has been reported,⁷ but the authors state that their phosphorylation and deblocking reactions are unsatisfactory. One of the problems concerning the synthesis of such molecules is that phosphorylation of vic-diol systems using conventional P^V phosphorylating agents often leads to formation of undesired cyclic 5-membered phosphates. We suggest an efficient phosphorylation procedure for *myo*-inositol derivatives based upon phosphite chemistry using the reagent developed by Sinha et al.⁸ for DNA synthesis, 2-cyanoethyl N,N-di-isopropylphosphoramidochloridite (3), recently used for the synthesis of α -L-fucopyranosyl phosphates⁹ and branched ribonucleotides.¹⁰ Bisphosphitylation of protected adenosine using a different reagent has also been observed recently.¹¹ Our preliminary experiments revealed that a simple vic-diol such as propane-1,2-diol could be phosphitylated with (3) and subsequently converted into the bisphosphate in ca. 80% yield. We now report a synthesis of (±)-myo-inositol 4,5bisphosphate¹² (Scheme 1) using this approach. We chose this system since trans-vicinal 4,5-bisphosphate substitution is known to be a requisite structural feature for calcium mobilising activity in an inositol phosphate.²





Scheme 1. Reagents and conditions: i, (3), EtNPri₂ in CH₂Cl₂; ii, tetrazole, HOCH2CH2CN-CH2Cl2; iii, ButOOH; iv, NH4OH, 60 °C, then Na-liq. NH₃.

Thus, (\pm) -1,2,3,4-tetra-O-benzyl myo-inositol (4)¹³ was treated with (3) (2 equiv.) in dichloromethane to give the bisphosphoramidite (5). Treatment of (5) with tetrazole and 2-cyanoethanol gave the phosphite triester (6). ³¹P N.m.r. spectroscopy showed that about 90% of the phosphorus could be accounted for by an AB system assignable to the 4,5-bisphosphite triester system (6) (δ 140.45, 140.75 p.p.m.; ⁵J_{PP} 2.93 Hz) (Figure 1). To the best of our knowledge, apart from the exceptional case of interbenzylic long range phosphorus-phosphorus coupling through a π -system,¹⁴ no examples of such a coupling through five bonds in an aliphatic system have previously been reported,¹⁵ and the observation of this effect is a key indication that bisphosphitylation has occurred. This effect disappears when the bisphosphite (6) is oxidised to the bisphosphate (7).

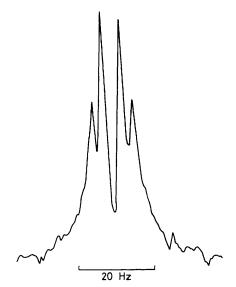


Figure 1. 24.15 MHz ³¹P n.m.r. spectrum of (±)-1,2,3,4-tetra-Obenzyl-myo-inositol-5,6-bisphosphite tetra-2-cyanoethyl ester (6) in CH₂Cl₂. ³¹P N.m.r. parameters were: sweep width 1 kHz; pulse width, 8 µs; recorded in 4K.

Oxidation of (6) with t-butyl hydroperoxide¹⁶ gave the phosphate triester (7). Ammonolysis removed the 2-cyanoethyl groups (heating to 60 °C for 3 h was necessary; mild ammonolysis removed only one 2-cyanoethyl group from each phosphotriester) and treatment with sodium in liquid ammonia¹⁷ quantitatively removed the benzyl groups to give (±)-myo-inositol 4,5-bisphosphate (8) [δ (pH 12) 4.68, 4.86 p.p.m.; ³J_{PH} 6.82, 7.30 Hz, respectively] in ca. 50% yield.

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